

## Supplemental Information for:

### Simulating the degree of oxidation in atmospheric organic particles

Heather Simon and Prakash V. Bhave

#### Derivation of equations used to treat POA oxidation in CMAQ

Dividing Equation (2) by OC (or the equivalent  $12 \times n_{OC}$ ) and combining with Equation (3), yields a relationship between OM/OC and  $\frac{n_O}{n_C}$  or  $\frac{n_H}{n_C}$ .

$$\frac{n_O}{n_C} = \frac{12}{15} \times \frac{OM}{OC} - \frac{14}{15} \quad \text{Equation (S1)}$$

$$\frac{n_H}{n_C} = \frac{44}{15} - \frac{12}{15} \times \frac{OM}{OC} \quad \text{Equation (S2)}$$

Dividing Equation (S2) by Equation (S1) yields a relationship between  $\frac{n_H}{n_O}$  and OM/OC,

$$\frac{n_H}{n_O} = \frac{\frac{44}{15} - \frac{12}{15} \frac{OM}{OC}}{\frac{12}{15} \frac{OM}{OC} - \frac{14}{15}} \quad \text{Equation (4b)}$$

which can be used in an NCOM mass balance to determine moles of oxygen.

$$NCOM = n_H + 16n_O = n_O \frac{n_H}{n_O} + 16n_O \quad \text{Equation (S3)}$$

$$n_O = \frac{NCOM}{16 + \frac{n_H}{n_O}} \quad \text{Equation (4a)}$$

Equations (S1) and (S2) allow the determination of  $\frac{n_O}{n_C}$  and  $\frac{n_H}{n_C}$  without having to explicitly track oxygen and hydrogen species in the model. In addition, these relationships can be used to convert measured OM/OC ratios to  $\frac{n_O}{n_C}$  and  $\frac{n_H}{n_C}$ . Equation (S1) can also be rewritten so that OM/OC is a function of  $\frac{n_O}{n_C}$ :

$$\frac{OM}{OC} = 1.25 \frac{n_O}{n_C} + 1.167 \quad \text{Equation (S4)}$$

Other studies have fit empirical relationships between OM/OC and  $\frac{n_O}{n_C}$ . Aiken et al.<sup>1</sup> found that data from downtown Mexico City and chamber measurements of SOA and POA fit the equation,

$$\frac{OM}{OC} = 1.26 \frac{n_O}{n_C} + 1.18, \text{ with an } R^2 \text{ of } 0.997. \text{ Sun et al. } ^2 \text{ found that data from Whistler Mountain, British}$$

Columbia fit the equation,  $\frac{OM}{OC} = 1.33 \frac{n_O}{n_C} + 1.18$ , with an  $R^2$  of 0.995. The similarity between these

empirical relationships and Equation (S4) support our use of Equation (S4) in Section 3.3 of this paper. Note that the data of Sun et al.<sup>2</sup> were not used by Heald et al.<sup>3</sup> when deriving Equation (2).

Table S1: Measured  $\frac{n_O}{n_C}$  and  $\frac{n_H}{n_C}$  ratios for OM sources and ambient OM.

OM type	$\frac{n_O}{n_C}$	$\frac{n_H}{n_C}$	Reference
Mobile source POA	0.03-0.05		1, 4
Plastic combustion POA	0.03-0.05		1, 4
Cooking POA	0.1	1.8-1.9	4, 5
Biomass combustion POA	0.3-0.45	1.45-1.8	1, 4, 5
SOA	0.3-0.4	1.5	1, 5
Mexico City fresh urban OM	0.5		1, 6, 7
Aged OM from Mexico City	0.6-0.8		1, 6, 7
Mexico City biomass burning OM	0.3-0.8		6, 7
Mexico City regional OM	0.7		1
Whistler Mountain aged OM	0.9-1.1		2
Whistler Mountain fresher OM	0.6-0.7		2
Pacific NW – younger Asian Layer	0.6		8
Pacific NW – older Asian layer	0.8		8
California central valley	0.5		8

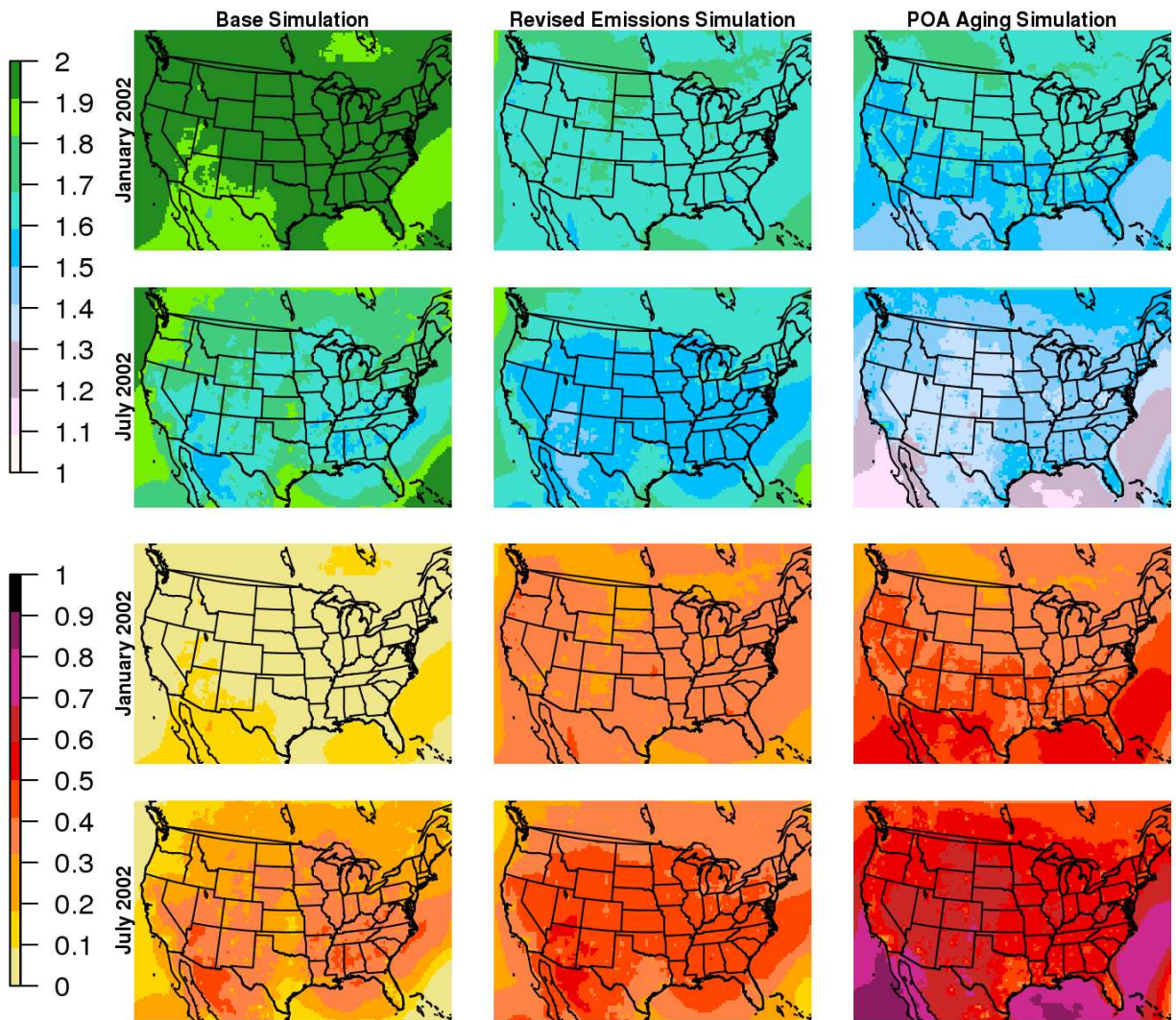


Figure S1. Monthly average  $\frac{n_H}{n_C}$  ratios for January 2002 (top),  $\frac{n_H}{n_C}$  ratios for July 2002 (2<sup>nd</sup> from top),  $\frac{n_O}{n_C}$  ratios for January 2002 (2<sup>nd</sup> from bottom), and  $\frac{n_O}{n_C}$  ratios for July 2002 (bottom). Results are presented for the basecase simulation (left), the revised emissions simulation (middle) and in the revised emissions + POA aging simulation (right).

## References

1. Aiken, A. C.; Decarlo, P. F.; Kroll, J. H.; Worsnop, D. R.; Huffman, J. A.; Docherty, K. S.; Ulbrich, I. M.; Mohr, C.; Kimmel, J. R.; Sueper, D., et al., O/C and OM/OC ratios of primary, secondary, and ambient organic aerosols with high-resolution time-of-flight aerosol mass spectrometry. *Environ. Sci. Technol.* **2008**, *42* (12), 4478-4485.
2. Sun, Y.; Zhang, Q.; Macdonald, A. M.; Hayden, K.; Li, S. M.; Liggio, J.; Liu, P. S. K.; Anlauf, K. G.; Leaitch, W. R.; Steffen, A., et al., Size-resolved aerosol chemistry on Whistler Mountain, Canada with a high-resolution aerosol mass spectrometer during INTEX-B. *Atmos. Chem. Phys.* **2009**, *9* (9), 3095-3111.
3. Heald, C. L.; Kroll, J. H.; Jimenez, J. L.; Docherty, K. S.; DeCarlo, P. F.; Aiken, A. C.; Chen, Q.; Martin, S. T.; Farmer, D. K.; Artaxo, P., A simplified description of the evolution of organic aerosol composition in the atmosphere. *Geophys. Res. Lett.* **2010**, *37*.
4. Mohr, C.; Huffman, J. A.; Cubison, M. J.; Aiken, A. C.; Docherty, K. S.; Kimmel, J. R.; Ulbrich, I. M.; Hannigan, M.; Jimenez, J. L., Characterization of Primary Organic Aerosol Emissions from Meat Cooking, Trash Burning, and Motor Vehicles with High-Resolution Aerosol Mass Spectrometry and Comparison with Ambient and Chamber Observations. *Environ. Sci. Technol.* **2009**, *43* (7), 2443-2449.
5. Huffman, J. A.; Docherty, K. S.; Mohr, C.; Cubison, M. J.; Ulbrich, I. M.; Ziemann, P. J.; Onasch, T. B.; Jimenez, J. L., Chemically-Resolved Volatility Measurements of Organic Aerosol from Different Sources. *Environ. Sci. Technol.* **2009**, *43* (14), 5351-5357.
6. DeCarlo, P. F.; Ulbrich, I. M.; Crounse, J.; de Foy, B.; Dunlea, E. J.; Aiken, A. C.; Knapp, D.; Weinheimer, A. J.; Campos, T.; Wennberg, P. O.; Jimenez, J. L., Investigation of the sources and processing of organic aerosol over the Central Mexican Plateau from aircraft measurements during MILAGRO. *Atmos. Chem. Phys.* **2010**, *10* (12), 5257-5280.
7. DeCarlo, P. F.; Dunlea, E. J.; Kimmel, J. R.; Aiken, A. C.; Sueper, D.; Crounse, J.; Wennberg, P. O.; Emmons, L.; Shinozuka, Y.; Clarke, A.; et al., Fast airborne aerosol size and chemistry measurements above Mexico City and Central Mexico during the MILAGRO campaign. *Atmos. Chem. Phys.* **2008**, *8* (14), 4027-4048.
8. Dunlea, E. J.; DeCarlo, P. F.; Aiken, A. C.; Kimmel, J. R.; Peltier, R. E.; Weber, R. J.; Tomlinson, J.; Collins, D. R.; Shinozuka, Y.; McNaughton, C. S.; et al., Evolution of Asian aerosols during transpacific transport in INTEX-B. *Atmos. Chem. Phys.* **2009**, *9* (19), 7257-7287.